

## Heteroepitaxy of Wide Bandgap Ternary Semiconductors

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## 1. Introduction

Ternary compounds encompass a large variety of structures with interesting magnetic, optical and electrical properties. The most thoroughly studied group of ternary semi-conductors are the I-III-VI<sub>2</sub> and II-IV-V<sub>2</sub> compounds that crystallize in the chalcopyrite (cp) structure (space group I42d). General incentives for the development of these compounds are their steep absorption edges, and the relatively high carrier mobilities as compared to other transition metal containing compounds, which, for example, may be utilized to advantage in thin film solar cell applications. Also, the valence band degeneracy that exists in the zb structure binary parent compounds is lifted due to the reduced symmetry of the cp structure compounds, providing potential advantages in the context of spin-polarized electron emission. Furthermore, some of the I-III-VI<sub>2</sub> and II-IV-V<sub>2</sub> compounds possess considerable birefringence and large second order susceptibility tensor components. Therefore, they are attractive for non-linear optical applications, such as harmonic generation and frequency mixing, extending the use of available lasers toward both higher or lower ranges of energy.

Since the cp structure is related to the zincblende (zb) structure (space group F43m) of the II-VI and III-V compounds by ordered substitutions on their cation sub-lattices, and the zb structure, in turn, is related to the diamond (space group Fd3m) structure of the group IV semiconductors by substitutions on the two fcc sublattices, there exist nearly matched equivalent *a*-axis lattice parameters\* for many of the cp structure ternary compounds and their zb/diamond structure isoelectronic analogs, as illustrated in Fig. 1. Thus large group IV, III-V and/or II-VI substrates are available for the heteroepitaxial growth of many cp structure materials. For example, the *a*-axis lattice parameters of ZnGeP<sub>2</sub> and ZnSiP<sub>2</sub> straddle the lattice constants of silicon and GaP, respectively (see Fig. 1). Therefore, the ZnSiP<sub>2</sub>-ZnGeP<sub>2</sub> system is suitable for providing wide bandgap ( $\geq 2.3$  eV) components in semiconductor heterostructures built on

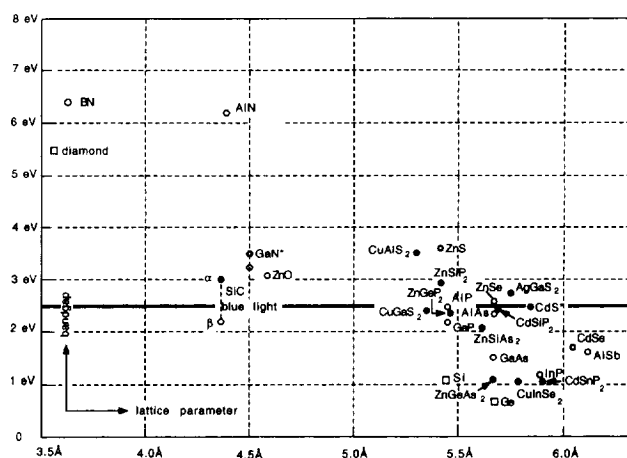


Fig. 1. Bandgaps and equivalent *a*-axis lattice parameters of selected ternary compounds and of their binary and diamond structure analogs. □ Diamond structure, ○ zinc blende structure, ○ wurtzite structure and ● chalcopyrite structure.

GaP or silicon substrates. In the context of non-linear optics, the access to heteroepitaxial structures, employing large diameter substrate wafers, provides for considerably larger beam paths than available with bulk single crystals. In addition, epitaxial growth methods achieve purer and more perfect materials than methods of high temperature bulk crystal growth and are suitable for improving the control of the stoichiometry and residual absorption in the transparency range. Furthermore, double and multiple heterostructures of the cp structure ternary compounds allow the engineering of the band offsets, refractive index profiles and strain. The utility of these properties for the realization of novel devices/circuits explains the current interest in the heteroepitaxial growth of ternary and multinary compounds.

Tables I and II list a variety of examples of epitaxial ternary compounds and alloys thereof reported during the past decade. The methods employed in this work are liquid phase epitaxy (LPE), flash evaporation (FE), metalorganic chemical vapor deposition (MOCVD), closed-tube halogen transport (CTHT), open-tube halide transport (OTHT) and molecular beam epitaxy (MBE). They are reviewed in more detail in §2 of this paper. The earlier literature is covered in the book by Shay and Wernick.<sup>1)</sup>

\*The equivalent *a*-axis lattice parameters listed for wurtzite structure materials equal  $\sqrt{2}$  times the *a*-axis lattice parameter of the hexagonal unit cell.

Table I. Selected I-III-VI<sub>2</sub> and II-IV-V<sub>2</sub> epilayers grown in the past decade.

Compound	Epitaxial methods <sup>a)</sup>	Reference
CuGaS <sub>2</sub>	LPE	Yamamoto <i>et al.</i> <sup>5)</sup>
CuInS <sub>2</sub> /GaP	LPE	Chang <i>et al.</i> <sup>6)</sup>
CuGaSe <sub>2</sub> , AgGaSe <sub>2</sub>	FE	Schumann <i>et al.</i> , <sup>8)</sup> Tempel <i>et al.</i> <sup>7)</sup>
LiInSe <sub>2</sub>	FE	Tempel <i>et al.</i> <sup>10)</sup>
CuGaS <sub>2</sub>	MOCVD	Hara <i>et al.</i> <sup>13)</sup>
CuAlS <sub>2</sub> , CuGaSe <sub>2</sub>	MOCVD	Hara <i>et al.</i> <sup>14)</sup>
AgGaS <sub>2</sub>	CTHT	Noda <i>et al.</i> <sup>30)</sup>
CuGaS <sub>2</sub>	OTHT	Yamauchi <i>et al.</i> <sup>31)</sup>
CuInSe <sub>2</sub>	MOCVD/MBE/LPE	Sato <i>et al.</i> <sup>27)</sup>
Cu <sub>5</sub> AlSe <sub>4</sub> <sup>b)</sup>	MBE	Morita and Narusawa <sup>26)</sup>
ZnGeP <sub>2</sub>	CTHT	Kataev <i>et al.</i> <sup>32)</sup>
ZnGeP <sub>2</sub>	MOCVD	Xing <i>et al.</i> <sup>16)</sup>
ZnGeAs <sub>2</sub>	MOCVD	Solomon <i>et al.</i> <sup>17)</sup>
ZnSnP <sub>2</sub>	LPE	Davis <i>et al.</i> <sup>4)</sup>

<sup>a)</sup>The abbreviations for the various epitaxial methods are explained in the text.

<sup>b)</sup>Tentative stoichiometry given by the authors.

Table II. Selected I-III-VI<sub>2</sub> alloy and II-IV-V<sub>2</sub> alloy epilayers grown in the past decade.

Alloys	Epitaxial methods	Reference
CuAl <sub>x</sub> Ga <sub>1-x</sub> S <sub>2</sub> , CuGaS <sub>2</sub> Se <sub>2-2y</sub>	MOCVD	Hara <i>et al.</i> <sup>14)</sup>
Li <sub>x</sub> Cu <sub>1-x</sub> InSe <sub>2</sub>	FE	Mitaray <i>et al.</i> <sup>11)</sup>
(ZnGeAs <sub>2</sub> ) <sub>x</sub> Ge <sub>1-x</sub>	MBE	Chelluri <i>et al.</i> <sup>24)</sup>
(ZnGeAs <sub>2</sub> ) <sub>x</sub> Ge <sub>1-x</sub>	MOCVD	Solomon <i>et al.</i> <sup>17)</sup>
(ZnGeP <sub>2</sub> ) <sub>x</sub> Ge <sub>1-x</sub>	MOCVD	Xiang <i>et al.</i> <sup>16)</sup>
ZnSi <sub>x</sub> Ge <sub>1-x</sub> P	MOCVD	Xing <i>et al.</i> <sup>16)</sup>

For a recent general review of the cp structure semiconductors the reader is referred to ref. 2.

## 2. Specific Heteroepitaxial Growth Methods

Liquid phase epitaxy has played an important role in the early exploration of heteroepitaxial structures of cp structure ternary semiconductors. For example, nearly lattice-matched CdSnP<sub>2</sub>/InP light emitting diodes and detectors were prepared in the early 1970s by LPE from tin solutions of cadmium and phosphorus below the peritectic temperature<sup>3)</sup> to provide light sources and detectors for optical communications in the wavelength region near 1.05  $\mu\text{m}$ , where silica-based optical fibers had minimum loss at this time. Recently, LPE has been used for the growth of ZnSnP<sub>2</sub> on GaAs.<sup>4)</sup> It has been applied successfully also to the growth of CuGaS<sub>2</sub><sup>5)</sup> and CuInS<sub>2</sub><sup>6)</sup> on GaP. However, seed dissolution, changing the liquidus and solidus compositions in the early stages of nucleation and growth, and the formation of defects that result in band tailing, are general problems of LPE of cp structure materials, which is suited primarily for applications that demand thick films without stringent control of their composition.

Insights into the epitaxial relation between I-III-VI<sub>2</sub> compounds and III-V substrates have been provided by studies of flash evaporated (FE) films of CuInSe<sub>2</sub>,<sup>7)</sup> CuGaSe<sub>2</sub>,<sup>8)</sup> AgGaSe<sub>2</sub><sup>9)</sup> on (001) GaAs and GaP sub-

strates. For the compounds with  $c/a$  ratios close to 2, the orientation of the epitaxial films is found to minimize the misfit strain at the interface and a wide temperature range of epitaxial growth exists even for substantially mismatched lattice constants. For AgGaSe<sub>2</sub> ( $c/a=1.82$ ), a narrower temperature range of epitaxial growth is observed. In the upper temperature range of epitaxial growth, the film is oriented relative to the substrate orientation minimizing the interfacial strain. However, a more complicated matching of planes and directions that are inclined to the surface normal at the lower limit of the temperature range of epitaxial growth. Investigations of LiCuSe<sub>2</sub> film growth on (001) GaAs and GaP substrates<sup>10)</sup> result in an even more complex behavior. LiInSe<sub>2</sub> crystallizes in bulk form in the  $\beta$ -NaFeO<sub>2</sub> structure (space group Pna2<sub>1</sub>) and is characterized by substantial differences in the cation radii. An analysis of reflection high-energy electron diffraction patterns shows the nucleation of cp structure LiInSe<sub>2</sub> that is overgrown by the rhombic structure as the epitaxial film thickens. Minimization of the interfacial misfit strain is no longer the dominant driving force that determines the relative orientations of the films, that is, Coulomb interactions become more important as the ionicity of the bonding increases. Alloys of composition Li<sub>x</sub>Cu<sub>1-x</sub>InSe<sub>2</sub> have been made that span the range of energy gaps between 1.05 and 2.06 eV with allowed direct transitions for  $x \leq 0.6$ .<sup>11)</sup>

Recently a relatively large number of papers has focused onto metalorganic chemical vapor deposition of both I-III-VI<sub>2</sub> and II-IV-V<sub>2</sub> compounds. Hara *et al.*<sup>12)</sup> reported in 1987 the successful growth of CuGaS<sub>2</sub> films on GaP(001) substrates, using cyclopentadienyl(triethylphosphine)copper(I), triethylgallium and hydrogen sulfide as source materials. This work was more recently expanded by the addition of triethyl-aluminum and hydrogen selenide sources to heterostructures employing CuGaSe<sub>2</sub> and CuAlS<sub>2</sub> as well as alloys of compositions CuGaS<sub>2-y</sub>Se<sub>2-y</sub> and CuAl<sub>x</sub>Ga<sub>1-x</sub>S<sub>2</sub>.<sup>13)</sup> Edge emission at 3.385 eV has been reported for CuAlS<sub>2</sub> epilayers on GaP. Also a narrowing of the FWHM of the luminescence for CuGaS<sub>1.30</sub>Se<sub>0.70</sub> as compared to CuGaS<sub>2</sub> has been reported. A similar narrowing of the FWHM of the luminescence has been observed by the author and his coworkers<sup>14)</sup> for bulk crystals of CuInS<sub>2-y</sub>Se<sub>2-y</sub> at  $y \approx 1$  as compared to pure CuInS<sub>2</sub> and CuInSe<sub>2</sub> crystals. The reasons for this behavior, which is contrary to the expected broadening of the luminescence due to alloy scattering, are presently not fully understood and warrant further investigations.

Also, substantial progress has been made with regard to the epitaxial growth of ZnGeP<sub>2</sub><sup>16)</sup> and ZnGeAs<sub>2</sub><sup>17)</sup> by MOCVD on GaP(001) and GaAs(001) substrate wafers, respectively, using dimethylzinc, germane, phosphine and arsine in a hydrogen carrier as source materials. For comparable conditions, the epitaxial growth of ZnGeP<sub>2</sub> on GaP(111) resulted in a smaller growth rate and larger density of microstructural defects than growth on the GaP(001) surface. With proper flow conditions, epitaxial layers of ZnGeP<sub>2</sub> as well as metastable alloys close to the pseudobinary ZnGeP<sub>2</sub>-Ge (i.e., of approximate composition Zn<sub>x</sub>GeP<sub>2-x</sub> with  $0 \leq x \leq 1$ ) were grown on GaP(001)

with mirror smooth surface morphology and excellent microstructural properties.

Figure 2 shows a cross sectional electron micrograph of a GaP/ZnGeP<sub>2</sub>/GaP heterostructure, which is the first III-V/II-IV-V<sub>2</sub> double heterostructure reported<sup>18)</sup> (see also ref. 2). Also, multiple heterostructures of GaP and ZnGeP<sub>2</sub> have been made with excellent control of the interface morphology by MOCVD. Current efforts in the author's laboratory focus onto the optimization of the resolution of II-IV-V<sub>2</sub>/III-V multiple heterostructures by chemical beam epitaxy and investigations in the ZnSi<sub>x</sub>Ge<sub>1-x</sub>P<sub>2</sub> system.

At constant phosphine and dimethylzinc flow rates  $F[\text{PH}_3]$  and  $F[\text{Zn}(\text{CH}_3)_2]$ , respectively, the Ge content in the alloys, and also the *c*-axis lattice parameter of the alloys decrease with decreasing germane flow rate  $F[\text{GeH}_4]$ . Figure 3 shows a plot of the *c*-lattice parameter change as function of flow rate ratio  $R = F[\text{Zn}(\text{CH}_3)_2]/F[\text{GeH}_4]$  for  $0.85 \leq x \leq 1$ . It shows that the *c*-axis lattice parameters of the films decrease continuously towards the *c*-axis lattice parameter of bulk single crystals of ZnGeP<sub>2</sub>. Note that even at the highest values of *R* the *c*-axis lattice parameter of the epilayers does not reach the bulk value  $c(\text{ZnGeP}_2) = 10.771 \text{ \AA}$ . The observed saturation of the *c*-axis lattice parameters of the epilayers at the value of  $10.792 \text{ \AA}$  is explained by elastic compression in the *a*- and *b*-lattice directions due to the larger than 1 ratio of the *a*-axis lattice parameters of ZnGeP<sub>2</sub> and GaP.<sup>19)</sup>

As a consequence of the cation ordering, superlattice reflection spots appear at [010] zone axis diffraction pattern of ZnGeP<sub>2</sub> epitaxial films of nearly stoichiometric composition. For alloys containing an excess of Ge the composition varies close to, but not necessarily exactly on the pseudobinary Ge-ZnGeP<sub>2</sub> of general composition Zn<sub>x</sub>GeP<sub>2x</sub> ( $0 \leq x \leq 1$ ). As *x* decreases, these superlattice reflections gradually weaken, vanishing into the background at  $x \approx 0.95$ . Therefore, for alloys with  $x \leq 0.95$  the electron diffraction pattern appears to be zincblende. However, although the *c*-axis lattice parameter of the epitaxial layers is larger than that of bulk ZnGeP<sub>2</sub> and increases with increasing Ge concentration in the films, as indicated by X-ray diffraction data, they remain smaller than two times the *a*-axis lattice parameter of ZnGeP<sub>2</sub> for all alloys studied at present. Thus, the tetragonal distortion persists to alloy compositions  $x \leq 0.90$ , so that partial ordering is maintained even for heavily Ge substituted alloys. In contrast, even for a relatively small excess of Ge in bulk single crystals, grown by directional solidification of a nearly stoichiometric melt, Ge precipitates are observed, while no precipitation of a second phase is observed for alloys obtained by MOCVD. Therefore, we conclude that observed solid solutions exist in metastable state.<sup>20)</sup>

As expected for a continuous range of solid solutions, the bandgap exhibits a monotonic narrowing with decreasing *x*. Since the tetrahedral covalent radius of Ge is too large for interstitial incorporation into the ZnGeP<sub>2</sub> lattice, the transition from chalcopyrite structure ZnGeP<sub>2</sub> towards the diamond structure lattice of Ge upon alloying proceeds probably via substitutions, that is, GeZn and GeP antisite formation. In the absence of direct

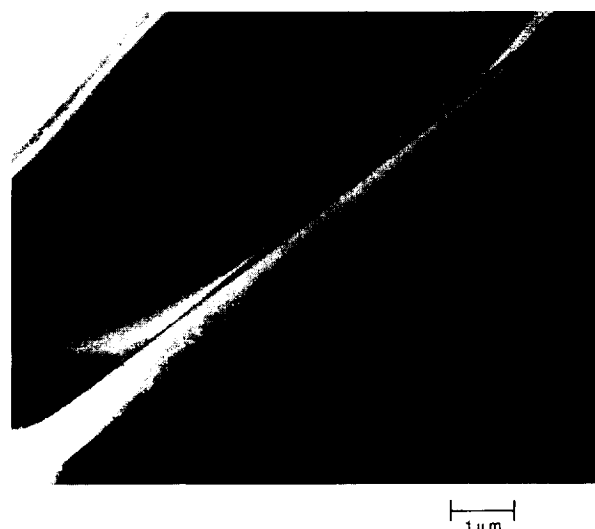


Fig. 2. Cross sectional transmission electron microscopy image of a GaP/ZnGeP<sub>2</sub>/GaP double heterostructure.

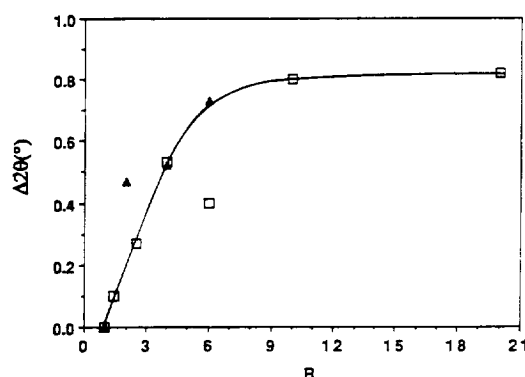


Fig. 3. Lattice parameter of excess germanium containing ZnGeP<sub>2</sub> alloys formed by MOCVD versus the ratio *R* of the dimethylzinc and germane flow rates.

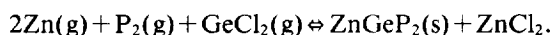
proof, and in view of the incomplete knowledge concerning the phase relations in the Zn-Ge-P system, further studies are needed to corroborate this conclusion. In the case of ZnGeAs<sub>2</sub>, investigations by Borshchevskii *et al.*<sup>21)</sup> and Schön *et al.*<sup>22)</sup> revealed an extended range of solid solution Ge<sub>x</sub>(ZnAs<sub>2</sub>)<sub>1-x</sub> for  $x > 0.5$ , which were interpreted as an equilibrium property of the Zn-Ge-As system. Thus the tendency to accommodate excess Ge seems to be enhanced for the arsenide as compared to the phosphide, which may be expected due to the positions of P and As relative to Zn and Ge in the periodic table. Note that the maximum melting point of the zinc-germanium pnictides are probably not located on the pseudobinaries ZnY<sub>2</sub>-Ge (Y=P, As). This is an additional reason why expressing the excess Ge containing solid solutions as Ge<sub>x</sub>(ZnY<sub>2</sub>)<sub>1-x</sub> represents a simplification. A distinct difference in the behavior of ZnGeP<sub>2</sub> and ZnGeAs<sub>2</sub> epilayers on III-V substrates has been observed also with respect to the formation of antiphase domain boundaries, which are much more prevalent in the arsenide.<sup>23)</sup> The cause for this is not understood and deserves further attention.

MBE of ZnGeAs<sub>2</sub> on GaAs has been reported in 1987

by Chelluri *et al.*,<sup>24)</sup> but is limited by the low sticking coefficient of, Zn at high temperatures and the formation of  $\text{Zn}_3\text{As}_2$  and  $\text{ZnAs}_2$  at low temperatures. This leaves a very narrow process window for  $\text{ZnGeAs}_2$  growth in the vicinity of  $360^\circ\text{C}$ , resulting Ge-rich alloys of  $\text{ZnGeAs}_2$ . The formation MBE of I-III-VI<sub>2</sub> compounds has been pioneered by Kazmerski<sup>25)</sup> and has been applied more recently by Morita and Narusawa<sup>26)</sup> and Sato *et al.*<sup>27)</sup> The work of Morita and Narusawa resulted in the epitaxial growth on GaAs(001) of a new phase  $\text{Cu}_5\text{AlSe}_4$ <sup>28)</sup> that have n-type conductivity, but can be type-converted by annealing at  $500^\circ\text{C}$ .<sup>29)</sup> Since  $\text{Cu}_5\text{AlSe}_4$  has a bandgap of  $\sim 3.1\text{ eV}$ <sup>29)</sup> it is a new wide bandgap material, exhibiting upon photoexcitation broad blue emission.

In the context of in-plane second harmonic generation and frequency mixing of infrared laser radiation, it is necessary to produce epitaxial films of several micrometer thickness and high optical quality. MBE and CBE are not fit for this task because of their notoriously small growth rates. Higher density nutrients are needed for enhancing the growth rate without the penalty of defect formation. Both physical and chemical transport processes lend themselves to achieving this goal. Closed-tube chemical vapor transport has been used successfully by for the growth of  $\text{AgGaS}_2$  crystals that exhibit upon photoexcitation at 4.2 K blue emission at 2.699 and 2.695 eV, corresponding to the free exciton and a neutral donor-bound exciton, respectively.<sup>30)</sup> Both closed-tube iodine transport and open tube halide transport using,  $\text{CuCl}$ ,  $\text{GaCl}_3$  and  $\text{H}_2\text{S}$  have been used to grow bulk crystals and epilayers of  $\text{CuGaS}_2$  on GaAs(001), respectively.<sup>31)</sup> Photo-excitation at 77 K results in addition to weak excitonic emission (FE: 2.5004 eV) in prominent donor-acceptor emission at 2.31 eV. Utilizing this luminescence peak as a probe for the evaluation of the uniformity of the film properties yields encouraging results.

Closed-tube halide transport has been used for the homoepitaxial growth of  $\text{ZnGeP}_2$ .<sup>32)</sup> A thermodynamic analysis of the concentrations of vapor species reveals that in the range  $873 \leq T \leq 1273\text{ K}$  the most prominent vapor species are  $\text{ZnCl}_2$ ,  $\text{Zn}$ ,  $\text{GeCl}_2$ ,  $\text{P}_2$  and  $\text{P}_4$ , that is, the transport occurs dissociatively according to the overall reaction



However, excess phosphorus additions were found to be essential for producing homo-epitaxy with excellent surface morphology on (100) and (001), but a pyramidal growth morphology on (112) substrates.

High pressure vapor transport (HPVT) of  $\text{ZnGeP}_2$  in a dense phosphorus vapor atmosphere has been reported by Xing *et al.*<sup>33)</sup> In view of the required temperature gradient between the source and the substrate, convective flow is inevitable under the conditions of HPVT. Figure 4 shows the results of the modeling of the gas flow dynamics and heat transport under the conditions of HPVT which give valuable information concerning the choices that need to be made in the thermal boundary conditions to optimize the vapor transport at the source and to establish uniform flow at the surface of the substrate. The calculation is based on a finite-element program (FIDAP)

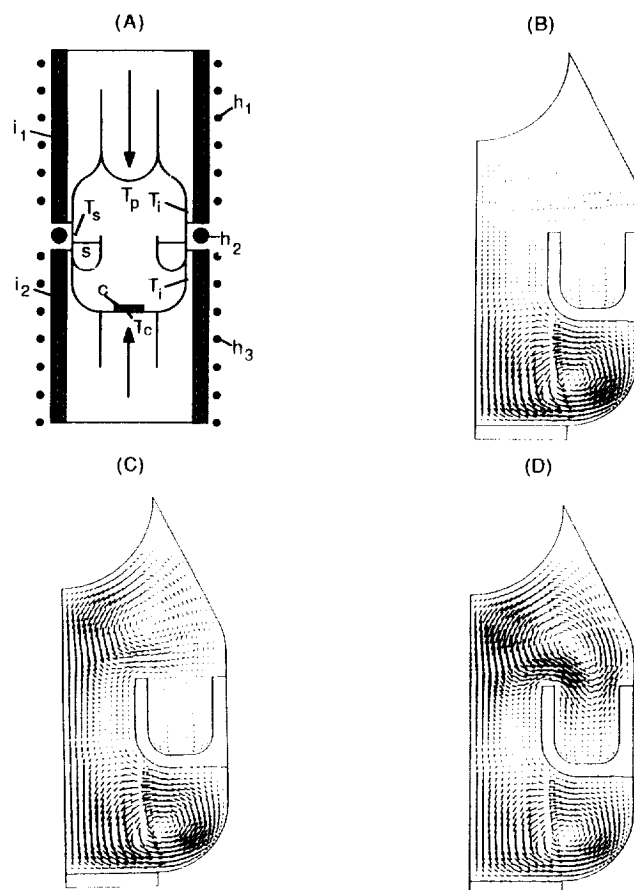


Fig. 4. (A) Schematic representation of an ampoule used for high pressure vapor transport:  $h_1$ ,  $h_2$ ,  $h_3$  independently controlled heaters,  $i_1$ ,  $i_2$  isothermal furnace liners, p, w locations of heat extraction by gas jets impinging onto the outside of the capsule. (B)–(D) Flow contours for 1 atm  $\text{P}_2 + \text{P}_4$  vapor representing the flow velocities and directions by arrows. Boundary conditions: (B)  $T_p = T_i = T_s = 1298\text{ K}$ ,  $T_w = 1223\text{ K}$ ; (C)  $T_s = T_i = 1298\text{ K}$ ,  $T_p = 1288\text{ K}$ ,  $T_w = 1223\text{ K}$ ; (D)  $T_p = 1288\text{ K}$ ,  $T_i = 1298\text{ K}$ ,  $T_s = 1348\text{ K}$ ,  $T_w = 1223\text{ K}$ .

that solves the Navier-Stokes equations, including temperature dependent chemical equilibria, for the ampoule geometry shown in Fig. 4(A), which is similar to the one introduced by Scholz.<sup>34)</sup> The thermal boundary conditions are controlled by a separately controlled heating elements ( $h_1 - h_3$ ) and the extraction of heat by gas jets impinging onto the outside of the fused silica envelope both on the top plug (p, temperature  $T_p$ ) and on the bottom fused silica window (w, temperature  $T_w$ ) that supports the growing crystal (c). The heater  $h_2$  establishes a spike  $T_s$  in the temperature at the location of the source (s). Two independently heated isothermal furnace liners ( $i_1$  and  $i_2$ ) keep the outer portions of the ampoule above and below  $h_2$  at a constant temperature  $T_i$ . The substrate is located on the fused silica window above A. Figures 4(B)–4(D) show the results of modeling of the vapor flow in 1 atm of a  $\text{P}_2/\text{P}_4$  atmosphere, revealing the turning-on of convective mixing over the annular source region and downward laminar flow over the substrate surface located on w. Thus far our HPVT work has been limited to self-seeded platelets, which are of n-type conductivity and exhibit a lower residual absorption in the transparency range than  $\text{ZnGeP}_2$  crystals grown from the melt.<sup>33)</sup>

In the opinion of the author, an important future topic of the epitaxial growth of cp structure materials will be heterostructures employing silicon as the substrate. Research in this field will be of interest in the context of integrating silicon technology with optical electronics as well as special applications, such as the use of polysilicon as a substrate material for thin film cp structure materials with applications in solar energy conversion. There are two approaches to implement heterostructures of Si and cp structure materials: 1) the growth of the cp structure epilayers on III-V buffer layers on Si and 2) the direct growth of graded films of selected cp structure materials on Si.

In the author's laboratory research is presently being conducted with the aim at nearly lattice-matched GaP epilayers on Si which are suitable substrates for the subsequent growth of either silicon or  $\text{ZnSi}_x\text{Ge}_{1-x}\text{P}_2$ . The former is being explored also in the context of dielectric isolation of Si by GaP. Highly selective GaP epitaxy has been achieved on patterned partially  $\text{SiO}_2$  covered hydrogen-terminated  $\text{Si}(001) 1 \times 1$  surfaces by CBE at  $310^\circ\text{C}$ , using triethyl gallium and tertiarybutylphosphine as source materials.<sup>35)</sup> Although the backbonding of silicon surface atoms in the  $\text{SiO}_2$  covered areas of the surface prevents the dative bonding of the t-butPH<sub>2</sub> and provides thus for excellent chemical selectivity, which is desirable in the context of microelectronics applications, the chemical selectivity provided by the ease of dative bonding to Ga surface atoms of nucleated GaP islands on Si as compared to silicon surface atoms is a nuisance since it favors three-dimensional growth. Therefore, a molecular layer epitaxy step is essential for sealing the silicon surface. Another problem of GaP epitaxy by CBE is the high carbon doping that leads to low resistivity p-type films. Therefore, alternative low thermal budget methods of carbon-free growth are of considerable interest, that is, warrant the exploration of halide transport.

On the other hand, we have shown by MOCVD that  $\text{ZnSi}_x\text{Ge}_{1-x}\text{P}_2$  films, which also closely match the lattice constant of silicon, can be produced with substantially higher resistivity than pure  $\text{ZnGeP}_2$  or GaP films.<sup>36)</sup> Furthermore the exact lattice-matching on a plane extending from the silicon corner of the composition tetrahedron Ge-Si-ZnGeP<sub>2</sub>-ZnSiP<sub>2</sub> to points on the Ge-ZnSiP<sub>2</sub> and ZnGeP<sub>2</sub>-ZnSiP<sub>2</sub> pseudo-binaries that match the lattice constant of silicon, provides for another incentive to explore the direct graded growth of exactly lattice-matched metastable quaternary alloys. The control of such a knitting process preserving exact lattice matching, charge neutrality and structural integrity during grading from pure Si into metastable quaternary alloys all the way to a terminal layer of  $\text{ZnSi}_{0.5}\text{Ge}_{0.5}\text{P}_2$  represents a substantial challenge, pushing the limits of current epitaxial technologies. We have chosen CBE to meet this challenge.

### 3. Conclusions

The past decade has brought a significant increase in the research effort devoted to heteroepitaxy of ternary cp structure compounds and alloys. Thus far this effort has focused at III-V compound substrates, but is likely to include in future increasingly research on silicon- and, to a

lesser extent, germanium-based heterostructures. Both double and multiple heterostructures employing nearly lattice-matched III-V/II-IV-V<sub>2</sub> combinations have been made with excellent interfacial properties so that the realization of confined heterostructures is bound to happen, in the near future. Amazingly well behaved defect structures with wide energy gaps have been discovered as a result of the present effort concerning the heteroepitaxy of ternary compound semiconductors and alloys. They warrant further exploration. In the opinion of the author, such an exploratory effort should include also a thorough evaluation of composite structures that may result in useful new materials.

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